

Comparison of the Aroma Composition and Sensory Properties of Dark Chocolates made with Moist Incubated and Fermented Cocoa Beans

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1 **Abstract**

2 In a previous investigation “moist incubation”, was described as a novel postharvest treatment for cocoa,
3 and the aroma composition of the resulting cocoa nibs was compared to unfermented and fermented
4 cocoa nibs. For this treatment unfermented and dried nibs are rehydrated with an aqueous solution
5 containing lactic acid and ethanol to adjust the pH-value and are subsequently incubated at 45°C under
6 aerobic conditions for 72 h before drying. The aim of the present study was to investigate the sensory
7 properties and aroma composition of dark chocolates made of these materials after roasting. Therefore,
8 gas-chromatography-olfactometry (GC-O) in combination with aroma extract dilution analysis (AEDA),
9 quantitation with isotopically labelled standards, odor activity value (OAV) determination and sensory
10 analysis were performed. The three different chocolates had distinct sensory and OAV profiles. The
11 sensory profiles showed a higher intensity of fruity aroma notes and lower intensity of bitterness and
12 astringency in the chocolate made with the moist incubated cocoa, while the chocolate made of fermented
13 cocoa reached higher scores in the roasty aroma notes. Furthermore, higher OAVs were determined for
14 the Strecker aldehydes in the chocolate made of the moist incubated cocoa, whereas higher OAVs for the
15 pyrazines and the acids were detected in the chocolate made of fermented cocoa. In contrast, the
16 chocolate produced with the unfermented cocoa showed low cocoa specific aroma notes and high levels
17 of astringency and bitterness. The detected differences reveal interesting insights into the influence of
18 different postharvest treatments on the resulting aroma composition in the final chocolate. Furthermore,
19 the alternative postharvest treatment was demonstrated to result in chocolates with a pleasant sensory
20 profile.

21 **Keywords**

22 Cocoa postharvest treatment; cocoa incubation; dark chocolate; cocoa aroma formation; sensory
23 evaluation

24

25 **Introduction**

26 Cocoa is the main ingredient for chocolate, being one of the most favored sweets worldwide, beloved for
27 its very distinct sensory properties. Before fresh cocoa beans can provide a desired aroma as well as the
28 favored slight bitterness and astringency, biochemical transformation within the fresh beans' chemical
29 composition is needed. During the traditional postharvest treatment, a spontaneous microbial degradation
30 of the adhering fruit pulp surrounding the beans, leads to conditions inducing the desired biochemical
31 changes in the beans.^{1,2} The key factors can be summarized as the acidification of the beans' tissue,
32 targeting a pH value of approximately 4.5-5.5, a temperature rise to approximately 45-50 °C, and the
33 availability of oxygen.²

34 The reconstruction of the traditional fermentation process under controlled conditions in vitro without
35 the influence of microorganisms was subject of many previous studies.³⁻⁷ For this "fermentation-like
36 incubation" beans were removed from the fresh cocoa pod, depulped and then incubated at controlled
37 temperatures in pH-adjusted solutions. It was shown that the formation of aroma-relevant precursors as
38 well as a directed transformation of polyphenols⁵ could be achieved to the same extent as during
39 traditional fermentations so a possible commercial use was discussed.⁴ However, this process is restricted
40 to the use of fresh beans, hence the process has to take place on or close to a farm site. Furthermore,
41 rather high expenditures for infrastructure are needed. Therefore, an alternative approach, independent
42 from time- and location referred to as "moist incubation" has been proposed.⁸ In contrast to the
43 fermentation-like incubation using beans freshly removed from the cocoa pod, for the moist-incubation
44 unfermented and dried cocoa nibs are used, which are storable and may be transported to any production
45 site. For the treatment unfermented and dried nibs are rehydrated with an aqueous solution containing
46 lactic acid and ethanol to adjust the pH-value and are subsequently incubated at 45°C under aerobic
47 conditions for 72 h before drying. During a first investigation,⁸ the aroma formation before and after this
48 treatment was investigated on a molecular level and compared to fermented cocoa. The results indicated

49 that aroma formation within the beans can be achieved independently of microbial degradation of the
50 pulp, when applying the moist incubation treatment on unfermented and dried beans. However, the
51 results showed differences in the abundance of certain cocoa key odorants. Esters and Strecker aldehydes
52 were found in equal or higher quantities in the moist incubated sample compared to the fermented sample.
53 On the other hand, the fermented sample showed higher quantities in compounds such as acetic acid and
54 2- and 3-methylbutanoic acid. The material was investigated after applying the postharvest treatment
55 including a drying step to directly compare the effect of the applied postharvest treatment without further
56 processing like roasting. However, it remained unclear, if the detected differences are still detectable in
57 the final product, the chocolate.

58 Therefore, the aim of the present study was to characterize the sensory properties and decode the aroma
59 profiles on a molecular level of the same materials used in the previous study (moist incubated,
60 fermented, and unfermented cocoa)⁸ after processing them into model chocolates. Gas-chromatography-
61 olfactometry (GC-O) in combination with aroma extract dilution analysis (AEDA), quantitation with
62 isotopically labelled standards and sensory analysis⁹ were performed to decode the aroma properties of
63 the three model chocolates on molecular level and gain further insights on the influence of the different
64 postharvest treatments on the generation of cocoa key odorants in the final products, the chocolates.

65

66 **Material and Methods**

67 **Chemicals**

68 For identification and determination of retention indices, the following chemicals were used: acetic acid,
69 (*E,E*)-2,4-decadienal, 2,3-diethyl-5-methylpyrazine, dimethyl trisulfide, 2-ethyl-3,5-dimethylpyrazine,
70 ethyl 2-methylbutanoate, ethyl 3-methylbutanoate, ethyl methylpropanoate, 2-ethyl-5-methylpyrazine,
71 ethyl 3-phenylprop-2-enoate, ethyl 3-phenylpropanoate, 3-ethylphenol, ethyl phenylacetate, 3-hydroxy-
72 4,5-dimethylfuran-2(*5H*)-one, 4-hydroxy-2,5-dimethyl-3(*2H*)-furanone, 2-isobutyl-3-methoxypyrazine,

73 2-methoxyphenol, 2- and 3-methylbutanal, 2- and 3-methylbutanoic acid, 2-methyl-
74 3(methyldithio)furane, methylpropanoic acid, (*E,E*)-2,4-nonadienal, phenylacetic acid, 2-phenylethanol,
75 2-phenylethyl acetate, and trimethylpyrazine were purchased from Sigma Aldrich Chemie GmbH
76 (Buchs, Switzerland).

77 For quantitation, the following isotopically substituted standards were used: 2-(²H₃)methylbutanal, 3-
78 (²H₃)methyl(3,4,4,4-²H₄)butanal, ethyl 3-(²H₃)methyl(2,2,3,4,4,4-²H₆)butanoate, (²H₆)dimethyl
79 trisulfide, 2-(³H₂)methyl-3,5-dimethylpyrazine, 2-(²H₅)ethyl-3,6-dimethylpyrazine, 2-(1,1-²H₂)ethyl-
80 3(1,1-²H₂)ethyl-5-(²H₃)methylpyrazine, 2-methyl-3-((²H₃)methyldithio)furane, 3-(²H₃)methyl-
81 (2,2,3,4,4,4-²H₆)butanoic acid, 2-(²H₅)phenylethyl acetate, (²H₅)ethyl-3-phenylpropanoate, 2-
82 (²H₅)phenylethanol, 4-hydroxy-2-methyl-5-(¹³C)methyl(5-¹³C)furan-3(2*H*)-one, ethyl-3-
83 (²H₅)phenyl(2,3-²H₂)prop-2-enoate, phenyl(¹³C₂)acetic acid (AromaLAB GmbH, Martinsried,
84 Germany), (¹³C₂)acetic acid (Merck KGaA, Darmstadt, Germany).

85 **Raw Materials**

86 The moist incubated and dried cocoa, the unfermented and dried, as well as the fermented and dried
87 cocoa material as obtained during the previous study⁸ were used to prepare the prototype chocolates.
88 Cocoa of the cultivar Trinitario was harvested on a farm in Costa Rica and a batch of approximately 800
89 kg was filled in a wooden fermentation box and covered with banana leaves to start the fermentation.
90 Mixing and aeration by transferring the mass to the next box was firstly performed after 48 h and was
91 repeated every 24 h until a total fermentation time of approximately 120 h was reached. The beans were
92 then spread on trays in a drying hall and dried under occasional mixing for approximately 10 days. To
93 obtain unfermented material, one part of the fresh beans was directly spread on wooden drying trays to
94 suppress fermentation and dried on trays in the same way as fermented cocoa beans. Samples of
95 fermented and unfermented beans were shipped to Switzerland and stored at 12 °C until they were broken
96 and deshelled to obtain unfermented and fermented cocoa nibs. For the moist incubated material eight

97 portions of 150 g (± 0.1 g) unfermented nibs were rehydrated under vacuum in a sealed bag for 12 h at 4
98 °C with 80 g (± 0.1 g) of aqueous solution containing lactic acid (0.1 mol/L) and ethanol (5 % v/v) to
99 reach a pH value in the cocoa solids of 5.1 and a final moisture content of 35 %. The bags were then
100 opened, fumigated with oxygen, sealed and then incubated at 45 °C for 72 h in a laboratory incubator
101 under occasional mixing by turning the bags every 12 h. After incubation the material was mixed and
102 dried on trays using a laboratory oven at 40 °C for 24 h in a laboratory oven with air circulation under
103 occasional turning until a final moisture content < 6 % was reached.

104 For the formulation of the chocolate prototypes commercially available deodorized cocoa butter (Carma,
105 Barry Callebaut AG, Zurich, Switzerland), white crystal sugar (Schweizer Zucker AG, Frauenfeld,
106 Switzerland), and sunflower-lecithin (Bunge Ltd., Chesterfield, USA) were used.

107 **Preparation of Prototype Chocolates**

108 Sample materials were frozen with liquid nitrogen (PanGas AG, Dagmersellen, Switzerland) and ground
109 with a kitchen blender (Thermomix[®], Vorwerk AG, Dierikon, Switzerland) to a particle size < 2 mm. To
110 provide a reproducible, homogenous, quick roasting, and subsequent quick cooling of the cocoa material,
111 an adapted roasting method based on a thin layer roasting technique developed by Mohr (1970)¹⁰ was
112 performed: approximately 50 g of finely ground cocoa powder was evenly distributed with a maximum
113 layer thickness of 3 mm on one half of a 30 cm \times 60 cm sheet of aluminum foil and covered with the
114 other half after folding in the middle. The edges were folded to keep the cocoa powder in place. These
115 envelopes were then roasted for 10 min in an oven (type H 5081-60 BP, Miele AG, Spreitenbach,
116 Switzerland) between two preheated ($125 \text{ °C} \pm 2 \text{ °C}$) tailor-made solid aluminum plates (35 cm \times 30 cm
117 \times 1.5 cm) with a thermocouple (type T, EBI 40 TC-01, Xylem Analytics Germany Sales GmbH & Co.
118 KG, Ingolstadt, Germany) attached measuring the temperature of the plate at the point of contact with
119 the envelope. Pre-trials showed that the temperature of the cocoa powder inside the envelope reached the
120 temperature of the aluminum plates within 60 s. A roasting time of 10 minutes was defined in pre-trials,
121 in which this process time resulted in the most balanced aroma profile of the material. After the roasting,

122 the envelopes were removed and placed flat on a steel surface to quickly cool down to room temperature.
123 The roasted cocoa material was hereafter mixed with crystal sugar and refined two times with a 3-roll
124 refiner (Type SDY 200, Bühler AG, Uzwil, Switzerland) to reach a particle size below 25 μm . This
125 premix was manually homogenized with cocoa butter and lecithin to prepare a 70 % prototype chocolate,
126 containing 50 % cocoa mass, 20 % cocoa butter, 29.5 % sugar, and 0.5 % lecithin. The chocolate masses
127 were not conched to avoid further changes in the aroma constitution after roasting. Pre-trials showed that
128 good textural properties can be achieved by this preparation technique. The chocolates were then
129 manually pre-crystallized and filled in chocolate bar molds. After complete crystallization, the bars were
130 wrapped in aluminum foil, vacuum packed, and frozen at $-20\text{ }^{\circ}\text{C}$ until use for analysis. The chocolate
131 samples are hereafter referred to as “incubated chocolate”, “fermented chocolate”, and “unfermented
132 chocolate”.

133 **Methods**

134 **Sensory Analysis**

135 The sensory evaluation of the obtained chocolates was carried out in the form of profiling with a trained
136 panel (n=8), referring to the *Quantitative Descriptive Analysis* (QDA[®]) method, and according to the
137 ISO 13299:2016 standard. Altogether, 10 attributes in the three main categories aroma, taste, and texture
138 were defined (Table S1). For profiling of the chocolate samples, the intensity of the chosen attributes
139 was rated on a continuous line scale from “0 = not perceivable” to “10 = very intense”. These evaluations
140 were done in the sensory lab of the Zurich University of Applied Sciences (ZHAW), Wädenswil,
141 Switzerland. The samples were blinded by labeling with random three-digit codes. The experimental
142 design was set up according to a *Randomized Complete Block Design* (RCBD), meaning that each sample
143 was randomly assigned to each panelist and each panelist was then evaluating all three samples in one
144 single session. Panelists were invited for an additional session to do a second evaluation of all three test
145 samples. The presentation of the samples was carried out one by one, following a sequential-monadic
146 presentation order. For neutralization between the samples water and saltless crackers were used. The

147 data was analyzed using the statistical software XLSTAT 2018 (Addinsoft, New York, USA), carrying
148 out a two-way analysis of variance (ANOVA) and a post-hoc test (Fisher's L.S.D.) to determine
149 significant differences between the samples. The results of the evaluation and the corresponding standard
150 deviations can be found in Table S2.

151 **Sample Preparation and Isolation of Volatiles for GC-O Analysis and Quantitation**

152 Aroma compounds were isolated in the same manner as previously described.^{8,11} To prepare an extract
153 for the GC-O analysis and identification of aroma compounds 20 g of chocolate were cut into fine pieces
154 with a kitchen knife and extracted with 200 ml diethyl ether by stirring at room temperature for 12 h. For
155 the quantitation of compounds in high and low concentrations samples of 2 g and 50 g were extracted
156 with 20 mL and 200 mL respectively in the same manner, after isotopically labelled standards of the
157 target compounds were added. Separation of the volatiles from the non-volatiles for the GC-O extract as
158 well as the extracts used for quantitation was performed using a SAFE distillation unit and the extract
159 was subsequently concentrated to a final volume of 300 μ L.

160 **Identification of Aroma Compounds Using Gas Chromatography-Olfactometry (GC-O) and Gas** 161 **Chromatography-Mass Spectrometry (GC-MS)**

162 GC-O in combination with AEDA and identification with GC-MS of selected compounds was performed
163 in the same manner, using the same equipment, as previously described.^{8,11,12}

164 **Quantitation of Selected Aroma Compounds**

165 The quantitation of selected compounds was done in the same manner, using the same equipment, as
166 previously described.⁸ The concentration of target compounds was calculated using a five-point
167 calibration line. To obtain the calibration lines, mixtures of analytes and isotopically substituted standards
168 in five different ratios (1:5, 1:2; 1:1, 2:1; 5:1) were analyzed. The peak area ratios of selected ions of
169 standard and analytes were plotted against the ratios of the respective concentrations. Quantitation of
170 analytes in the samples were determined with the calibration line using linear regression. The ions used
171 for quantitation, and the calibration lines can be found in the supporting information Table S3. All

172 samples were analyzed in triplicates (unless stated differently in the results table) and the results were
173 calculated as mean values.

174 **Results and Discussion**

175 **Sensory Profiles of the Dark Chocolates Made with Incubated, Fermented, and Unfermented** 176 **Chocolates**

177 The sensory scores in the defined attributes and the illustrated sensory profiles are shown in Figure 1.
178 While the incubated and the fermented chocolates were described having a differing, but typical dark
179 chocolate flavor profile with the pleasant attributes of dark chocolate such as malty, roasty, fruity, and
180 flowery aroma notes, as well as slight bitterness and low astringency, the unfermented chocolate was not
181 perceived as typical. The data showed a significant difference ($\alpha=0.05$) in many of the attributes between
182 the three samples (Table S1). The incubated chocolate showed a somewhat higher intensity score among
183 flowery, fruity, malty, and caramel-like aroma notes. On the other hand, roasty aroma notes were rated
184 higher in the fermented chocolate. The unfermented chocolate was rated with an overall low aroma
185 intensity with the highest score for the attribute green. Furthermore, the samples showed differences in
186 the perception of the taste attributes. The incubated chocolate was perceived sweeter than the other
187 chocolates. Additionally, the bitterness and astringency were perceived in a lower intensity compared to
188 the other samples. Among the three samples, the unfermented chocolate reached the highest scores for
189 both attributes, suggesting the desired transformation of polyphenols, usually induced by the
190 fermentation, and drying, was suppressed compared to the moist incubation and fermentation.

191 **Identification of Odor-Active Constituents in the Incubated, Fermented, and Unfermented** 192 **Chocolates**

193 Table 1 shows the results of the performed AEDA. Overall, 29 compounds with a flavor dilution factor
194 (FD factor) >4 have been detected in the investigated chocolates: 26 compounds in the incubated
195 chocolate sample, 25 compounds in the fermented chocolate, and 15 compounds in the unfermented

196 chocolate, respectively. One aroma compound with a FD factor >4 could not be identified by the
197 identification criteria mentioned in Table 1.

198 Many well-known aroma compounds which have also been previously found in roasted cocoa and dark
199 chocolates such as 2- and 3- methylbutanal, 2-ethyl-3,6-dimethylpyrazine, acetic acid, 2-isobutyl-3-
200 methoxypyrazine, 2-methyl-3(methyldithio)furan, 2- and 3-methylbutanoic acid, 2-phenylethyl acetate,
201 ethyl 3-phenylpropionate, 2-phenylethanol, 4-hydroxy-2,5-dimethyl-3(2*H*)-furanone, ethyl 3-phenyl-
202 prop-2-enoate, 3-hydroxy-4,5-dimethylfuran-2(5*H*)-one, and phenylacetic acid^{11,13} were detected in all
203 samples analyzed during this investigation. Additionally, an unknown compound with a meaty and nutty
204 odor was detected with high FD factor of 256 in the incubated chocolate and a FD factor of 128 in the
205 fermented chocolate This unknown compound was also detected with a somewhat lower intensity in the
206 unfermented chocolate with a FD factor of 16.

207 Overall, the fermented and incubated chocolates showed a comparable number of typical dark chocolate
208 and cocoa key aroma compounds, with comparable FD factors. As expected, fewer compounds with
209 generally lower FD factors were detected in the unfermented chocolate, indicating a higher concentration
210 of aroma precursors present in the material after fermentation or moist incubation.

211 The highest FD factor of 1024 was found for the caramel-like 4-hydroxy-2,5-dimethyl-3(2*H*)-furanone
212 in both the fermented and the incubated chocolates. This fact fits well with previous findings, where 4-
213 hydroxy-2,5-dimethyl-3(2*H*)-furanone was identified as one of the key odorants in the cocoa mass after
214 roasting and also in chocolate.^{11,14-16}

215 A major difference in the AEDA results can be seen for the earthy, nutty, and roasty smelling pyrazines.
216 2,3,5-Trimethylpyrazine, 2-ethyl-3,6-dimethylpyrazine, 2- ethyl 3,5-dimethylpyrazine, and 2,3-diethyl
217 5-methylpyrazine were detected with higher FD factors in the fermented chocolate compared to the
218 incubated chocolate, whereas only 2,3,5-trimethylpyrazine and 2-ethyl-3,6-dimethylpyrazine were
219 detected with a low FD factor of 4. In the past, many different pyrazines have been identified in the
220 aroma of roasted cocoa and chocolate and they are generally regarded as important contributors to the

221 aroma of chocolate.¹⁷⁻¹⁹ In contrast to that, a recent study where roasted Forastero beans were analyzed,
222 it was concluded that the pyrazines are of minor importance for the cocoa aroma.¹⁴

223 Furthermore, another difference between the aroma composition of the three chocolates of the present
224 study was detected among the organic acids, acetic acid, 2- and 3-methylbutanoic acid, and
225 methylpropanoic acid. These components showed higher FD factors in the fermented chocolate
226 compared to the incubated and the unfermented chocolate. Even though they are usually referred to
227 unpleasant odor descriptions such as pungent, sour, vinegar-like, sweaty, or rancid, in many studies these
228 acids reach the highest FD factors and consequently reach very high odor activity values in cocoa and
229 chocolate.^{13-16,20} The fact that many traditional chocolate making processing steps, like drying, roasting,
230 thin layer treatment of cocoa liquor, and conching aim at reducing these compounds,^{19,21,22} underlines
231 that these compounds, especially when present in high concentrations, are rather undesirable.

232 Furthermore, the important malty Strecker aldehydes 2- and 3-methylbutanal were detected with
233 comparable FD factors in the incubated chocolate and in the fermented chocolate, respectively. These
234 compounds can be directly linked to the presence of their parent amino acids leucine and isoleucine in
235 the raw material. These amino acids are known to be important precursors released during fermentation
236 within the bean.^{2,22} Only minor differences were detected in terms of FD factors between the fermented
237 and incubated chocolates for the fruity and flowery esters. While ethyl methylpropanoate, ethyl 2-
238 methylbutanoate, and ethyl 3-methylbutanoate, ethyl phenylacetate, and 2-phenylethyl acetate reached
239 rather low FD factors in the incubated and fermented chocolates, ethyl 3- phenylpropionate and ethyl 3-
240 phenylprop-2-enoate showed high FD factors. Furthermore, the flowery smelling alcohol 2-phenyl-
241 ethanol was detected in all three chocolates with a relatively high FD factor of 256 for the fermented and
242 incubated chocolate and FD 128 for the unfermented chocolate sample, respectively. Previous studies
243 showed that this odorant is already present in unfermented cocoa and concentrations do not change
244 significantly during roasting.¹⁴⁻¹⁶

245 Other well-known odorants of cocoa and chocolate such as the cabbage-like dimethyl trisulfide and the
246 cooked meat-like 2-methyl-3-(methylthio)furane showed slightly higher FD factors in the incubated
247 chocolate compared to the fermented chocolate. In the unfermented chocolate dimethyl trisulfide could
248 not be detected and 2-methyl-3-(methylthio)furane showed a lower FD factor in comparison to the
249 incubated and fermented chocolates. Other studies showed that these odorants are present in cocoa after
250 fermentation and increase during roasting.¹⁴⁻¹⁶ In a recent study, where two commercially available dark
251 chocolates with 90 % and 99 % cocoa content were investigated, dimethyl trisulfide reached the highest
252 FD factors during GC-O analysis and even showed the highest OAV during quantitation of both samples,
253 underlining the importance of these compounds for the cocoa aroma.¹³

254 Furthermore, the fatty and green components (*E,E*)-2,4-nonadienal and (*E,E*)-2,4-decadienal were
255 detected in the incubated chocolate with FD factors of 4 and 64 and in the fermented chocolate with
256 slightly higher FD factors of 64 and 128, respectively. These compounds have been identified in
257 chocolate and cocoa and are known to be thermally induced lipid oxidation products.^{23,24} However, both
258 compounds were not detected in the unfermented chocolate.

259 Another main difference in the GC-O profiles of the samples was found for the phenolic and animalic
260 smelling 3-ethylphenol and the smoky smelling 2-methoxyphenol. 3-Ethylphenol was exclusively
261 detected in the unfermented chocolate with a FD factor of 16. 2-Methoxyphenol was detected with a high
262 FD factor of 256 in the fermented chocolate. On the other hand, the incubated chocolate showed a low
263 FD factor of 4 and this odorant was not detectable in the unfermented chocolate. The same trend for this
264 compound was observed in the corresponding unroasted raw materials, suggesting that its formation is
265 strongly linked to microbial activity during fermentation.⁸

266 **Quantitation of Selected Aroma Compounds in the Investigated Chocolate Samples**

267 The results of the quantitation of the selected aroma compounds in the incubated, the fermented and the
268 unfermented chocolates are shown in Table 2. Among the quantitated volatiles, acetic acid was the most
269 abundant compound in all the samples. The lowest concentration was measured in the incubated

270 chocolate (23.4 mg/kg), followed by the unfermented chocolate (40.0 mg/kg). The fact that less acetic
271 acid was found in the incubated chocolate is in accordance with the values measured in the raw material
272 before roasting. This way, a loss of acetic acid during the drying step of this treatment procedure can be
273 suggested.⁸ The higher value for the fermented chocolate (55.7 mg/kg) is in accordance with another
274 study, where quantities of 53.7-87.7 mg/kg were found in commercially available dark chocolates with
275 cocoa contents ranging from 70-85 %.¹¹ Furthermore, higher concentrations of 2- and 3-methylbutanoic
276 acid were also measured in the fermented chocolate (1,760 µg/kg and 3,450 µg/kg) compared to the
277 incubated chocolate (418 µg/kg and 1,330 µg/kg). In other studies, concentrations ranged from 391-1,670
278 µg/kg for 2-methylbutanoic acid and 438-3,320 µg/kg for 3-methylbutanoic acid were measured,
279 showing the values obtained in the present study can be compared to the ones of commercially available
280 chocolates.^{11,13} For the unfermented chocolate, far lower concentrations for these compounds were
281 observed (135 µg/kg and 195 µg/kg). 2- and 3-Methylbutanoic acid are known to increase during
282 traditional fermentation, but they can also be generated during Strecker degradation from their
283 corresponding parent amino acids leucine and isoleucine during thermal treatment.^{1,21,25} A slight increase
284 after roasting of cocoa beans was detectable in different studies.¹⁴⁻¹⁶ Therefore, the high concentrations
285 in the fermented chocolate of the present study may derive from both processing steps – the fermentation
286 and the subsequent roasting –, while the moderate content of the incubated chocolate might be linked to
287 the formation of these compounds during the roasting process.

288 Interesting results were also found for the malty compounds 2- and 3-methylbutanal. These odorants
289 showed higher concentrations in the incubated chocolate (274 µg/kg and 916 µg/kg) compared to the
290 fermented chocolate (104 µg/kg and 587 µg/kg) and the unfermented chocolate (53.1 µg/kg and 208
291 µg/kg). The same trend was found when the raw material was analyzed before roasting and preparation
292 of chocolate. These important compounds are known to be formed by Strecker degradation from their
293 parent α -amino acids leucine and isoleucine.²⁴ Previous studies showed, that Strecker aldehydes can be
294 released upon contact with water in dry foods^{26,27} and also from fermented and dried, unroasted cocoa

295 beans after treatment with water.²⁸ The results of the present study suggest that the combined effect of
296 the moist treatment as well as the formation of aldehydes during drying and subsequent roasting could
297 lead to overall higher amounts in the incubated chocolate in comparison to the fermented chocolate.
298 Furthermore, the pyrazine concentrations of the incubated chocolate for 2-ethyl-3,5-dimethylpyrazine
299 (7.91 µg/kg), 2-ethyl-3,6-dimethylpyrazine (26.0 µg/kg), 2,3,5-trimethylpyrazine (7.11 µg/kg), and 2,3-
300 diethyl-5-methylpyrazine (0.23 µg/kg) were comparable to the concentrations found in the unfermented
301 chocolate (8.88 µg/kg, 15.7 µg/kg, 6.34 µg/kg, and 0.13 µg/kg), while the concentrations in the fermented
302 chocolate showed 5- to 20-fold higher values (111 µg/kg, 120 µg/kg, 136 µg/kg, 1.84 µg/kg). Pyrazines
303 are known to be formed from α -aminoketones during Strecker degradation in the Maillard reaction²⁹, but
304 it was shown by Scalone et al. (2015) that they may also derive from oligopeptides.³⁰ Furthermore, it was
305 shown that the use of oligopeptides as precursors promotes pyrazine formation compared to the use of
306 free amino acids in model systems.^{30,31} Short peptides have gained increasing attention as being
307 significantly responsible precursors for cocoa aroma formation. Recently, 34 Amadori and Heyns
308 compounds deriving from di- and tripeptides in fermented and dried cocoa have been identified for the
309 first time.³² Unfortunately, the authors did not investigate the volatile profiles deriving from these
310 Maillard reaction intermediates. However, their presence in fermented and dried cocoa and the findings
311 that oligopeptides are known to promote the formation of pyrazines suggests that these precursors may
312 have been formed to a larger extent during microbial fermentation compared to the moist incubation.
313 Furthermore, the results in a study from Zou et al. (2018) showed a promoted formation of Strecker
314 aldehydes when free amino acids were used as Maillard reaction precursors as compared to
315 oligopeptides.³¹ The higher quantities of pyrazines found in the fermented sample and the higher
316 quantities of Strecker aldehydes in the incubated sample could therefore indicate that the formation of
317 peptides was promoted during fermentation, while higher quantities of free amino acids were generated
318 after the moist incubation treatment. However, this has to be proven by the respective precursor
319 measurements in the different materials. A possible reason for the different precursor formation may be

320 due to differences in the pH value reached in the cotyledon during the postharvest treatment. It is known
321 that lower pH values reached during fermentation promote the formation of oligopeptides, while higher
322 pH values promote the formation of amino acids, especially the important precursors leucine, isoleucine,
323 valine and phenylalanine.³³

324 The floral smelling 2-phenylethanol was quantitated in comparable amounts in the incubated (1,880
325 µg/kg) and the unfermented chocolate (1,830 µg/kg), while lower amounts were found in the fermented
326 chocolate (1,530 µg/kg). This is in accordance with values obtained during a previous study, where lower
327 amounts were measured in the fermented cocoa beans.⁸ 2-Phenylethanol can be converted by
328 microorganisms to form phenylacetic acid and 2-phenylethyl acetate during fermentation.²² This is also
329 in line with the higher concentrations for the beeswax-like phenylacetic acid and 2-phenylethyl acetate
330 in the fermented sample compared to the incubated and the unfermented chocolates. Another ester which
331 was found with a slightly higher concentration in the fermented chocolate is ethyl 3-phenylprop-2-enoate
332 (64.0 µg/kg) compared to the incubated (48.3 µg/kg) and the unfermented chocolate (33.6 µg/kg). On the
333 other hand, the esters ethyl 3-methylbutanoate and ethyl 3-phenylpropanoate were measured in higher
334 concentrations in the incubated chocolate with 3.94 µg/kg and 10.0 µg/kg compared to 1.88 µg/kg and
335 3.46 µg/kg in the fermented and 0.56 µg/kg and 3.93 µg/kg in the unfermented chocolate, suggesting that
336 these esters may have been formed to a larger extent enzymatically in the cocoa bean material during the
337 moist incubation.

338 Another important odorant is the caramel-like 4-hydroxy-2,5-dimethyl-3(2*H*)-furanone. It reached the
339 highest FD factors in the AEDA with a comparable concentration in the incubated chocolate (504 µg/kg)
340 compared to the fermented chocolate (548 µg/kg), while the unfermented chocolate showed a much lower
341 concentration (72.7 µg/kg). These results are in line with concentrations given in the literature.^{11,14-16}

342 In addition to that, two sulfur containing odorants were quantitated: the cabbage-like dimethyl trisulfide
343 and the meaty and nutty 2-methyl-3(methyldithio)furanone. Both odorants have been identified in
344 fermented cocoa beans and an increase of their concentrations was shown during roasting.^{14,15} In the

345 incubated chocolate and the fermented chocolate of the present study, concentrations of dimethyl
346 trisulfide were comparable, reaching 6.31 $\mu\text{g}/\text{kg}$ and 4.45 $\mu\text{g}/\text{kg}$, respectively, being in line with values
347 given in literature.^{11,13-15} On the other hand, the concentration of this odorant was drastically lower in the
348 unfermented chocolate with 0.69 $\mu\text{g}/\text{kg}$, suggesting that the corresponding precursors were formed during
349 the moist incubation treatment and the fermentation in comparable intensities, while the necessary
350 precursors were missing in the unfermented chocolate.

351

352 **Comparison of the Calculated *Odor Activity Values***

353 The calculated OAVs of the odorants in the investigated chocolates are shown in Table 3. Eleven
354 compounds with an $\text{OAV} > 1$ were detected in the incubated chocolate, twelve in the fermented and eight
355 in the unfermented chocolate. Overall, the incubated and fermented chocolates reached higher values
356 compared to the unfermented chocolate. The values determined for the incubated chocolate were
357 somehow lower in comparison to the OAVs of the fermented chocolate. A major difference in the OAV
358 profile of these two chocolates is the 13-fold higher OAV for 2-ethyl-3,5-methylpyrazine in the
359 fermented chocolate, and the somewhat higher OAV determined for the Strecker aldehydes 2- and 3-
360 methylbutanal in the incubated chocolate.

361 The highest OAV in the incubated chocolate were observed for dimethyl trisulfide (210), 3-
362 methylbutanoic acid (120), and phenylacetic acid (114), followed by acetic acid (61), 3-methylbutanal
363 (61), and 4-hydroxy-2,5-dimethyl-3(2H)-furanone (19). On the other hand, the fermented chocolate
364 showed the highest values for 3-methylbutanoic acid (313), phenylacetic acid (162), and dimethyl
365 trisulfide (148), followed by acetic acid (145) and 2-ethyl-3,5-dimethylpyrazine (65). The highest OAV
366 in the unfermented chocolate was reached by acetic acid (104), followed by phenylacetic acid (18), and
367 3-methylbutanal (14).

368 The different OAVs for all odorants of the three chocolates are reflected in the differences found for their
369 sensory profiles. The high OAVs of pyrazines in the fermented sample might be linked to the intense

370 perception of the attribute roasty, and the lower scores in the fruity and flowery perception in comparison
371 to the incubated sample during sensory evaluation. However, the fact that the incubated and the
372 fermented chocolate both showed typical dark chocolate aroma properties despite the low amounts of
373 pyrazines in the incubated sample confirms the findings of Frauendorfer et al. (2019) that pyrazines are
374 of negligible importance for the cocoa aroma.¹⁴ Furthermore, a twofold higher OAV for the fruity ester
375 ethyl 3-methylbutanoate was determined for the incubated chocolate (4) in comparison to the fermented
376 chocolate (2). Even though these OAVs were both comparably low, it is possible that fruity and flowery
377 notes were suppressed by the stronger roasty aroma notes in the fermented chocolate, while the low
378 concentrations of pyrazines support the flowery and fruity aroma perception of the incubated chocolate.
379 A major difference in the aroma composition has been discovered among the pyrazines, which were
380 measured in much higher concentrations in the fermented chocolate compared to the incubated and
381 unfermented chocolate. This may be due to a promoted formation of oligopeptides during fermentation,
382 while the higher amounts of Strecker aldehydes in the incubated chocolate suggests a promoted formation
383 of free amino acids.

384 The aroma of all chocolates was perceived as different from each other, which is not surprising, because
385 of the given differences of the postharvest treatments used in comparison. Overall, the fermented
386 chocolate as well as the incubated chocolate showed a typical aroma and taste profile. In comparison, the
387 unfermented chocolate did not elicit the pleasant attributes of dark chocolate, such as malty, roasty, fruity,
388 and flowery aroma notes and was mostly perceived as green. Thereby, the green odor impression might
389 be a result of the low abundancies of 2- and 3-methylbutanal, the pyrazines, dimethyl trisulfide and 4-
390 hydroxy-2,5-dimethyl-3(2H)-furanone and dimethyl trisulfide and on the other hand the relatively high
391 odor activity value of acetic acid and also the presence of 2-isobutyl-3-methoxypyrazine, which
392 concentration and OAV was not determined due to its relatively low FD-factor. Furthermore, a high
393 adstringency and bitterness could be perceived in the unfermented chocolate. In this study, it was
394 observed that reduction of astringency and bitterness as reached during fermentation and drying, can also

395 be achieved by applying the moist incubation treatment. This might be linked to the extant polyphenol
396 oxidase activity of the incubated cocoa tissue in addition to the use of an oxygen atmosphere during the
397 incubation, as well as the contact of the incubated cocoa powder during drying. The present study showed
398 that the moist incubation treatment of unfermented and dried nibs provides an intermediate product,
399 which can be used to produce a chocolate with a pleasant aroma and taste. Thus, the proposed technique
400 has the potential to serve as an alternative reproducible time- and location-independent postharvest
401 treatment, which can be easily controlled. Besides, this study gives interesting insights about the
402 formation of typical cocoa aroma compounds, especially the Strecker aldehydes and pyrazines, which
403 were formed to different extents, depending on the applied postharvest treatment. To understand the
404 underlying mechanisms leading to the obtained aroma and taste profile, more research regarding the non-
405 volatile components is necessary.

406 **Abbreviations Used**

407 AEDA, Aroma Extract Dilution Analysis; FD, Flavor Dilution; FFAP, Free Fatty Acid Phase; GC-MS,
408 Gaschromatography-Mass Spectrometry; GC-O, Gaschromatography-Olfactometry; OAV, Odor
409 Activity Value; SAFE, Solvent Assisted Flavor Evaporation; ZHAW, Zurich University of Applied
410 Sciences

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414 **Conflict of Interest**

415 The authors declare no competing financial interest.

416 **Supporting Information Description**

417 **Table S1.** Attributes, Definitions and References for the Evaluation of the Incubated, Unfermented
418 and Fermented Chocolates

419 **Table S2.** Means Scores of Attributes, p-values from ANOVA and Results from Post-Hoc Test
420 (Fisher`s L.S.D.) from the Sensory Evaluation of the Incubated, Unfermented and
421 Fermented Chocolates

422 **Table S3.** Cocoa Odorants, Standards, Selected Ions (m/z) of Analytes, Standards and Calibration
423 Lines Used For Quantitation

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Figure Captions

Figure 1. Sensory Profiles of the Incubated (IC), Unfermented (UC), and Fermented Chocolate (FC)

Table 1 Flavor Dilution Factors of Compounds Determined in Aroma Distillates Isolated from Incubated Chocolate (IC), Unfermented Chocolate (UC) and Fermented Chocolate (FC) during AEDA

no. ^a	odorant ^b	odor quality ^c	retention index on			FD factor ^d	
			FFAP	OV-1701	IC	UC	FC
1	2- and 3-methylbutanal ^{e,f}	malty	927	710	256	16	128
2	ethyl methylpropanoate ^g	fruity	950	818	8	<4	<4
3	ethyl 2-methylbutanoate ^g	fruity	1019	907	8	<4	16
4	ethyl 3-methylbutanoate ^g	fruity	1042	910	8	<4	<4
5	dimethyl trisulfide ^g	cabbage-like	1358	1030	64	<4	16
6	2-ethyl 5-methylpyrazine ^g	earthy	1376	n.d.	4	<4	<4
7	trimethylpyrazine ^g	earthy	1391	1080	4	<4	64
8	2-isopropyl-3-methoxypyrazine ^g	bell pepper-like	1414	1139	4	<4	16
9	2-ethyl-3,6-dimethylpyrazine ^g	earthy	1430	1153	4	4	256
10	acetic acid ^h	pungent	1439	n.d.	64	16	256
11	2-ethyl-3,5-dimethylpyrazine ^g	earthy	1446	1160	<4	<4	16
12	2,3-diethyl-5-methylpyrazine ^g	earthy	1475	1220	<4	<4	64
13	2-isobutyl-3-methoxypyrazine ^g	bell pepper-like	1505	1239	4	16	16
14	methylpropanoic acid ^h	pungent, sweaty	1551	n.d.	4	<4	16
15	2-methyl-3(methyldithio)furane ^g	meaty, nutty	1649	1265	128	16	64
16	2- and 3-methylbutanoic acid ^h	pungent, sweaty	1653	n.d.	128	16	128
17	(<i>E,E</i>)-2,4-nonadienal ^g	fatty, green	1686	1350	8	<4	64
18	unknown	meaty, nutty	1714	n.d.	256	16	128
19	ethyl phenylacetate ^h	flowery, fruity	1769	1360	4	<4	16
20	(<i>E,E</i>)-2,4-decadienal ^h	fatty, green	1795	n.d.	64	<4	128

(table continues)

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Table 1 Flavor Dilution Factors of Compounds Determined in Aroma Distillates Isolated from the Incubated Chocolate (IC), Unfermented Chocolate (UC) and Fermented Chocolate (FC) (continued)

no. ^a	odorant ^b	odor quality ^c	retention index on		FD factor ^d		
			FFAP	OV-1701	IC	UC	FC
21	2-phenylethyl acetate ^h	dried fruits-like, flowery	1799	1477	4	4	16
22	2-methoxyphenol ^g	smoky	1849	1226	4	<4	256
23	ethyl 3-phenylpropionate ^g	dried fruits-like, flowery	1867	n.d.	256	128	128
24	2-phenylethanol ^h	flowery	1898	1283	256	128	256
25	4-hydroxy-2,5-dimethyl-3(2 <i>H</i>)-furanone ^h	caramel-like	2025	1240	1024	128	1024
26	ethyl 3-phenylprop-2-enoate ^g	fruity, cinnamon-like	2118	n.d.	128	128	128
27	3-ethylphenol ^g	phenolic, animalic	2170	n.d.	<4	16	<4
28	3-hydroxy-4,5-dimethylfuran-2(5 <i>H</i>)-one ^g	seasoning	2190	1347	16	16	16
29	phenylacetic acid ^g	beeswax-like	2546	n.d.	32	16	64

a) number of identified compound based on retention index on capillary column FFAP, b) odorant name, c) odor quality perceived at sniffing port, d) flavor dilution factor determined by AEDA on capillary FFAP, e) flavor dilution factor determined by AEDA on capillary OV-1701, f) identification based on retention index and odor quality of compound found in literature³⁴, g) identification by comparison of odor quality at sniffing port and retention index on FFAP with reference substance, h) identification by comparison of odor quality at sniffing port, mass spectrum and retention index on FFAP with reference substance

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Table 2 Results of the Quantitation of the Odorants in the Incubated Chocolate (IC), Unfermented Chocolate (UC) and Fermented Chocolate (FC)

Sample	IC		UC		FC	
	content ($\mu\text{g}/\text{kg}$)					
odorant ^a	mean	rel. SD ^b (%)	mean	rel. SD ^b (%)	mean	rel. SD ^b (%)
acetic acid	23400	3.3	40000	2.5	55700	2.5
2-methylbutanoic acid	418	1.4	134	4.4 ^c	1760	3.0
3-methylbutanoic acid	1330	1.8	189	1.0 ^c	3450	3.2
phenylacetic acid	2950	3.4	724	12.5	4210	1.7
2-phenylethanol	1880	0.5	1830	0.5	1530	0.3
2-methylbutanal	274	1.2	53.1	2.2 ^c	104	0.0
3-methylbutanal	916	3.0	208	10.9	587	8.0
ethyl 3-methylbutanoate	3.94	2.8	0.56	13.6 ^c	1.88	2.4
2-phenylethyl acetate	36.6	3.1	42.5	0.3 ^c	257	1.1
ethyl-3-phenylprop-2-enoate	48.3	0.4	33.6	4.4	64.0	4.1
ethyl-3-phenylpropanoate	10.0	1.2	3.93	1.3	3.46	2.4
4-hydroxy-2,5-dimethyl-3(2H)-furanone	504	1.9	72.7	1.1	548	4.1
2-ethyl-3,5-dimethylpyrazine	7.91	1.5	8.88	0.8	111	0.7
2-ethyl-3,6-dimethylpyrazine	26.0	0.3	15.7	1.0	120	0.8
2,3,5-trimethylpyrazine	7.11	4.7	6.34	0.5	136	0.7
2,3-diethyl-5-methylpyrazine	0.23	8.9	0.13	3.2	1.84	16.0
dimethyl trisulfide	6.31	2.2	0.69	11.0	4.45	2.5
2-methyl-3-(methylthio)-furan	0.29	4.1	0.24	6.7	0.21	8.8

a) odorant name, b) relative standard deviation was calculated from quantitative data obtained from three extractions of each sample c) relative standard deviation was calculated from quantitative data obtained from two extractions of sample

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Table 3 Odor Activity Values Calculated for the Incubated Chocolate (IC), Unfermented Chocolate (UC) and Fermented Chocolate (FC)

odorant ^a	odor threshold ^b [$\mu\text{g}/\text{kg}$]	IC	UC	FC
		OAV ^c		
dimethyl trisulfide	0.03	210	<1	148
3-methylbutanoic acid	11	121	17	314
phenylacetic acid	26	113	28	162
acetic acid	350 ^d	67	114	159
3-methylbutanal	15	61	14	39
4-hydroxy-2,5-dimethyl-3(2H)-furanone	27	19	3	20
2-methylbutanal	34	8	<1	5
2-phenylethanol	490	4	4	3
2-ethyl-3,5-dimethylpyrazine	1.7	5	4	65
ethyl 3-methylbutanoate	0.98	4	1	2
2-methylbutanoic acid	110	4	<1	16
2-ethyl-3,6-dimethylpyrazine	76	<1	<1	2

a) odorant name, b) orthonasal threshold value determined in oil according to reference³⁴, c) odor activity value calculated as ratio of amount in sample to threshold value determined in oil, d) orthonasal threshold value determined in oil according to reference³⁵

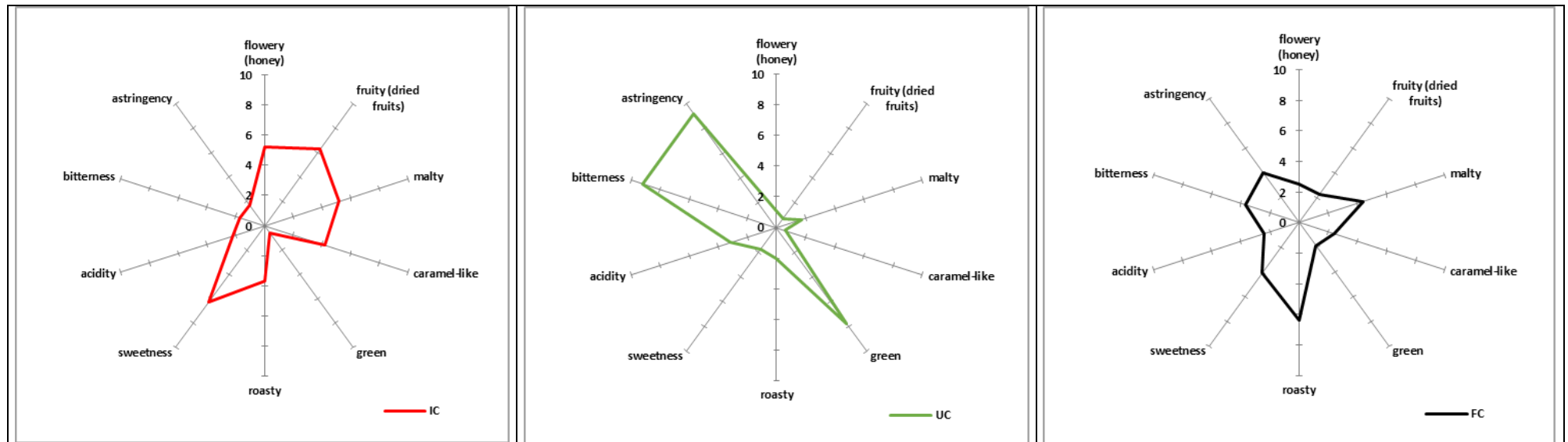


Figure 1 Sensory Profiles of the Incubated Chocolate (IC), Unfermented Chocolate (UC) and Fermented Chocolate (FC)

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